# SUBSTRATE TEMPERATURE AND ANNEALING EFFECTS ON THE STRUCTURAL AND OPTICAL PROPERTIES OF NANO-CdS FILMS DEPOSITED BY VACUUM FLASH EVAPORATION TECHNIQUE

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For deposition of CdS thin films was successfully used a vacuum flash evaporationtechnique.XRD results have shown that all CdS films deposited at substrate temperatures of  $100^{\circ}$ C,  $200^{\circ}$ C and  $300^{\circ}$ C exhibit a predominant sharp peak at  $2\theta = 26.5^{\circ}$  which assigned to the (002) plane of hexagonal structure. It was found that the intensity of diffraction peak (002) increases with a decrease of the substrate temperature. Average grain size calculated from XRD results is about 30 nm and weakly depends on the substrate temperature. After 30min annealing at  $400^{\circ}$ C the intensity of (002) peaks increased considerably and at the same time average grain size growth only on 5-10%. Profiler and AFM results have shown that the roughness and morphology of the CdS thin film surfaces vary insignificantly with the variation of substrate and annealing temperature. The transmission and reflection spectrum of CdS thin films measured in the range 400-1000nm. Optical measurements have shown that the films deposited at substrate temperatures of  $200^{\circ}$ C and  $300^{\circ}$ C have bandgap 2.42eV which corresponds to the bandgap of bulk CdS crystal. After annealing the bandgap values were changed negligible.

(Received June 8, 2016; Accepted August 1, 2016)

*Keywords:* Cadmium sulfide (CdS), Thin-film, Solar cell, Vacuum flash evaporation technique

#### **1. Introduction**

Cadmium sulfide (CdS) is a common material used in the formation of solar cells based on cadmium telluride (CdTe) and CuInGaSe<sub>2</sub> (CIGS). In such solar devices n-type CdS thin-film acts as an optical window due to high bandgap energy Eg=2.42eV. For solar cells application CdS films have to have relatively high conductivity to reduce electrical loses of solar cells, thin thickness to provide high transmission and good uniformity in order to prevent electrical short-circuit effect. The structural, optical and electrical properties of CdS thin films strongly depend on the applied technique and the substrate temperature [1]. Different techniques have been reported for the deposition of CdS thin films namely: vacuum thermal evaporation [2-6], spray pyrolysis [7], electro-deposition [8] and chemical bath deposition [9-11]. Among other techniques the vacuum evaporation technique is very simple, inexpensive and is well established technique for the preparation of large area uniform films. In contrast to many other techniques the process of vacuum evaporation does not lead to heating the substrate, so allows to control the substrate temperature accurately by the external heating. Due to this feature vacuum evaporation technique is applicable for formation of solar cells on a flexible polyimide substrate. Replacing the glass substrate to a flexible substrate reduces the weight of thin film solar cells by 98%. Such solar cells are very promising for application on spacecraft so far as they have high ratio of electric power to weight. Flexible solar cells represent also of interest in the market for the terrestrial applications as they can easily be mounted on the surface of various shapes.

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For fabrication of solar cell on flexible substrates it is necessary to use low-temperature film deposition techniques because polyimide substrates do not allow heating more than  $450^{\circ}$ C. As such techniques the most promising are the magnetron sputtering method and the method of vacuum thermal evaporation. For evaporation of multicomponent compound with different vapor pressure of components more suitable is the variant of vacuum thermal evaporation method notably the method of vacuum flash evaporation [12-17].

Distinguishing feature of this method is that the deposition of film is performed by continuous vacuum evaporation of small portions of multicomponent compound. Small particles of crushed multicomponent alloy whose constituents have different vapor pressure (for instance Cd and S in CdS) are falling into the preheated flash exchanger in a random way. Since the particles have different sizes and are falling in a random way in the flash exchanger at each moment there are some amount of particles which have different temperature and are at different evaporation stages of compound constituents. Simultaneous presence of such particles in the flash exchanger provides at the average the same fluxes of evaporated constituents and hence the stoichiometric composition of deposited film with pinpoints accuracy. Unfortunately as it is not surprising a very limited research works denoted to studies the characteristic of CdS films deposited by vacuum flash evaporation technique. This technique is also called a method of discrete evaporation.

The aim of this work was to study the structural, morphological and optical characteristics of nano-structured CdS thin films deposited on the glass substrates by flash evaporation technique as a transmittance and hetero layer in chalcogenide solar cells.

# 2. Experimental details

Thin films of CdS were fabricated by flash evaporation technique on commercial glass slides used as substrate with thickness 1 mm and diameter 20 mm. The substrates were thoroughly cleaned by soap-free detergent ultrasonically and then iteratively rinsing in distilled water to remove the traces of detergent. Thereafter the substrates were cleaned ultrasonically in ethanol and then dried by blowing air. The vacuum chamber before evaporation was pumped to a base pressure of  $1.5 \cdot 10^{-5}$  Torr.



Fig.1. Schematic representation of the experimental setup.

Schematic view of flash evaporation setup is presented in Fig. 1.The starting material was a pure bulk crystal of CdS which is crushed to a powder with particle dimensions of 50  $\mu$ m –150  $\mu$ m and then placed inside a batcher.

Dispensing the CdS particles from batcher onto the molybdenum flash exchanger was performed due to opening and closing of batcher outlet by the instrumentality of electro-vibrator (Fig.1).

The amount of CdS particles falling into the flash exchanger was controlled by opening time of outlet. The outlet is opened when electro-vibrator attracts up by electromagnet. In our

experiments the opening time was varied from 0.2 to 0.5 c. The outlet is opened again for evaporation of a next new portion of CdS particles only after evaporation of all particles contained in the flash exchanger. The evaporation process monitored through the chamber's window.

Dispensing the CdS particles was started after the flash exchanger

is heated up to temperature  $1200^{\circ}$ C -  $1300^{\circ}$ C suitable to evaporate the CdS powder. The distance between the flash exchanger and substrate was about 12 cm.

The film thickness and the duration of deposition were typically 200-300 nm and 20-40 min. The CdS films were deposited at substrate temperatures of  $100^{\circ}$ C,  $200^{\circ}$ C and  $300^{\circ}$ C. Asdeposited films were annealed at  $400^{\circ}$ C during 30 min in vacuum. The structural properties of samples were studied by X-ray diffractometer URD-6 in the  $\theta$ -2 $\theta$  mode using Cu - K<sub>a</sub> ( $\lambda$  = 1.5405 Å) radiation. The surface morphology and roughness of CdS films were investigated by ZYGO profiler and by atomic force microscopy (AFM)type NEXT supplied by NT-MDT Inc. Transmittance and reflectance, over the wavelength from 400 to 1000 nm were measured using double beam Filmetrics F20 spectrophotometer.

### **3. Results and Discussion**

### 3.1. Structural properties of as-deposited and annealed CdS thin films

Fig. 2 shows the X-ray diffraction patterns of as-deposited (red chart) and annealed (blue chart)CdS thin films deposited by flash evaporation technique on a glass substrate. Diffraction patterns were recorded over the  $2\theta$  range from  $15^0$  to  $60^0$  with the step  $0.1^0$ . XRD analyses showed that the CdS films were polycrystalline. Both as-deposited and annealed CdS films exhibit a predominant sharp peak at  $2\theta$  around  $26.5^0$  which can be assigned to either the (002) plane in the case of hexagonal structure [JCPDS card no: 41-1049] or the (111) plane in the case of cubic structure [JCPDS card no: 10-0454].

The exact interpretation of XRD patterns it is quite difficult because most peaks of cubic and hexagonal CdS structure differ negligible within in the very small angle.

The summery of XRD data of as-deposited CdS thin films at different substrate temperature are presented in Table 1. In Table 2 are presented the summery of XRD data after annealing.

It is seen from Table 1 that observed interplanar spacing-d and diffraction peaks in the XRD patterns when  $2\theta = 26.518^{\circ}$  (substrate temperature  $100^{\circ}$ C),  $2\theta = 26.591^{\circ}$ ,  $2\theta = 51.577^{\circ}$  (substrate temperature  $200^{\circ}$ C),  $2\theta = 26.473^{\circ}$ ,  $2\theta = 51.546^{\circ}$  (substrate temperature  $300^{\circ}$ C) correspond with reflection from the crystal planes (002), (112) of CdS hexagonal structure, which agrees well with standard XRD patterns (JCPDS card no: 41-1049). The same results of hexagonal structure are observed for CdS thin films deposited by flash evaporation technique in work [14].

After annealing, as can be seen from Table 2 the structure of the deposited films clearly changed. After the peaks around  $2\theta = 51^{\circ}$  (for CdS films as-deposited at  $200^{\circ}$ C and  $300^{\circ}$ C) disappeared and the predominant sharp peaks at  $2\theta$  around  $26.5^{\circ}$  drastically have increased (Fig. 2, blue charts).



Fig. 2. XRD patterns of CdS films as-deposited (red chart) at substrate temperature: a)  $300^{0}$ C, b)  $200^{0}$ C, c)  $100^{0}$ C and after annealed at  $400^{0}$ C during 30 min (blue chart)

Substrate	$2\theta, (^{0})$	d (Å)	d (Å)	[hkl] plane	c(Å)	c(Å)
temperature		observed	standard		observed	standard
		from XRD	JCPDS		values	JCPDS
$T = 100^{\circ}C$	26.518	3.3614	3.3567	002	6.7228	6.7134
T=200°C	26.591	3.3523	3.3567	002	6.7046	6.7134
	51.577	1.7721	1,761	112	Х	Х
T=300 <sup>0</sup> C	26.473	3.3670	3.3567	002	6.734	6.7134
	51.546	1.7731	1,761	112	Х	Х

Table 1. XRD data of as-deposited CdS thin films at different substrate temperatures

Table 2. XRD data of CdS thin films deposited at different substrate temperatures after annealing

Substrate	$2\theta, (^{0})$	d (Å)	d (Å)	d (Å) [hkl] plane		c(Å)
temperature		observed	standard		observed	standard
_		from XRD	JCPDS		values	JCPDS
$T = 100^{\circ}C$	26.541	3.3585	3.3567	002	6.7170	6.7134
$T=200^{0}C$	26.579	3.3537	3.3567	002	6.7074	6.7134
$T=300^{0}C$	26.524	3.3607	3.3567	002	6.7214	6.7134

The structural parameters of as-deposited and annealed CdS thin films such as the average grain sizes (D), dislocation density ( $\delta$ ), the number of crystallites per unit area (N) and microstrain ( $\epsilon$ ) for comparison are presented in Table 3.

Substrate temperature	2θ ( <sup>0</sup> )		(deg)		D (nm) from XRD results		[hkl] plane
	as- deposited	annealed	as- deposited	annealed	as-deposited	annealed	as- deposited/ annealed
T=100 <sup>0</sup> C	26.518	26.541	0.35	0.3	29.49	34.74	002
$T=200^{0}C$	26.591	26.579	0.35	0.35	30.43	30.27	002
T=300°C	26.473	26.524	0.35	0.35	28.96	29.57	002
	1		1				
Substrate temperature	$\frac{\delta}{10^{15}}$ lines/m <sup>2</sup>		ε x10 <sup>-4</sup>		$\frac{N}{x10^{15}}$ m <sup>-2</sup>		Thickness of films (nm) (t)
	as-deposited	annealed	as-deposited	annealed	as-deposited	annealed	as-deposited/ annealed
T=100 <sup>0</sup> C	1.15	0.83	11.8	10	8.576	5.246	220
$T=200^{0}C$	1.08	1.09	11.4	11.5	8.514	8.650	240
$T=300^{0}C$	12	1 14	12	11.7	12 353	11.6	300

*Table 3. Structural parameters of CdS films as-deposited at substrate temperature*  $100^{0}$ C,  $200^{0}$ C and  $300^{0}$ C and after annealing at  $400^{0}$ C during 30 min

The average grain sizes (D) of the film was calculated using the Scherrer's equation [18]:

$$D = \frac{0.9\lambda}{\beta \cos(\theta)}$$

where  $\lambda$  is the wavelength of the X-ray used ( $\lambda$ =1.5405 Å),  $\beta$  is the full-width at half-maximum (FWHM) of the (002) peak which has maximum intensity, and ( $\theta$ ) is the Bragg's angle.

The variation of the grain size with substrate temperature is presented in Table 3. It is seen that the grain size vary insignificantly. The biggest grain size (30.43 nm) observed for CdS film as-deposited at  $200^{\circ}$ C and (34.74 nm) for CdS film deposited at  $100^{\circ}$ C after annealing.

The dislocation density  $\delta$  defined as the length of dislocation lines per unit volume, has been calculated using the equation [19]:  $\delta = 1/D^2$ . The results of calculation are presented in Table 3. The small values of ( $\delta$ ) confirmed the good crystallinity of the CdS films. The number of crystallites per unit area (N) and the micro-strain ( $\epsilon$ ) of the films have been estimated with the use of the following equations [20]: N=t/D<sup>3</sup> and  $\epsilon = \beta \cos(\theta)/4$ . It is seen from Table 3 the annealing led to decreasing the micro-strains and number of crystallites per unit area of all as-deposited films. The lower values of micro-strains and the number of crystallites per unit area observed for as-deposited CdS films at 200°C and for CdS film deposited at 100°C after annealing.

### 3.2. Optical properties of as-deposited and annealed CdS thin films

Using transmittance (T) and reflectance (R) spectra the absorption coefficient has been calculated by the following equation:

$$\alpha = \frac{1}{t} ln \frac{1-R}{T}$$

where *t* is the thickness of CdS film.



Fig.3. Plots of  $(\alpha \hbar \omega)^2 vs.$  photon energy  $(\hbar \omega)$  of CdS thin films deposited at 100°C, 200°C and 300°C substrate temperatures.



Fig.4. Plots of  $(\alpha \hbar \omega)^2 vs.$  photon energy  $(\hbar \omega)$  of CdS thin films deposited at  $100^{\circ}C$ ,  $200^{\circ}C$  and  $300^{\circ}C$  substrate temperatures after annealing at  $400^{\circ}C$  during 30minin vacuum

Above the fundamental absorption edge the dependence of the absorption coefficient on the incident photon energy is given by Tauc's model [21]:

$$\alpha \hbar \omega = B(\hbar \omega - E_g)^n$$

where  $\hbar\omega$  is the photon energy,  $E_g$  is the optical bandgap, *B* is a constant and *n* is an exponent that depends on the type of optical transitions. For direct allowed transitions  $n = \frac{1}{2}$ . The variation of  $(\alpha\hbar\omega)^2$  of CdS thin films deposited at different substrate temperatures as a function of photon energy are shown in Fig.3. In Fig.4 are presented the variation of  $(\alpha\hbar\omega)^2$  vs. photon energy of CdS thin films after annealing.

Using the liner extrapolation method the value of the optical bandgap for as-deposited CdS thin films is determined to be 2.39, 2.42 and 2.42 eV for  $100^{\circ}$ C,  $200^{\circ}$ C and  $300^{\circ}$ C, accordingly. After annealing the optical bandgap is changed insignificantly and became 2.38, 2.40 and 2.40 eV, accordingly. The results of measurements have shown that the CdS thin film deposited at  $200^{\circ}$ C and  $300^{\circ}$ C have the same optical bandgap as the bulk of CdS crystal  $E_{g}$ = 2.42 eV.

#### 3.3. Morphological properties of as-deposited and annealed CdS thin films.

Fig. 5 shows typical 1,5µm x 1,5µm AFM images of CdS films deposited on a glass substrate at different temperatures by flash evaporation technique. For statistics a few AFM images of different sites of the surface of each sample were investigated. Profiler results and AFM

images show that the morphology and roughness of the surface weakly depends on the substrate temperature. A typical AFM images for as-deposited and annealed CdS thin films are presented in Fig.5 and Fig.6, respectively.



Fig.5. A typical AFM images and the distribution of grain size of as-deposited CdS thin films on a glass substrate at  $100^{\circ}C$ .



Fig.6. A typical AFM images and the distribution of grain size of CdS thin films deposited on a glass substrate at 100°C after annealing

# 4. Conclusion

The structural, morphological and optical properties of the CdS thin films deposited on glass substrate at different temperatures  $(100^{\circ}\text{C}, 200^{\circ}\text{C} \text{ and } 300^{\circ}\text{C})$  by vacuum flash evaporation technique have been investigated. The annealing effect on these properties has also been investigated. XRD results showed that both all of as-deposited at different temperatures and annealed CdS thin films were polycrystalline, have hexagonal structure and exhibit a predominant sharp (002) peak at 20 around 26.5°. It is found that the intensity of diffraction peak (002) increases with a decrease of the substrate temperature. After annealing the peak at  $2\theta = 51^{\circ}$  (for CdS films as-deposited at  $200^{\circ}\text{C}$  and  $300^{\circ}\text{C}$ ) disappeared and the predominant sharp peaks at  $2\theta = 26.5^{\circ}$  increased.

Using XRD results the structural parameters of CdS thin films such as the average grain sizes (D), dislocation density ( $\delta$ ), the number of crystallites per unit area (N) and micro-strain ( $\epsilon$ ) was calculated. These results have shown that the best crystalline structure have CdS films as-deposited at 200<sup>o</sup>C and CdS films deposited at 100<sup>o</sup>C after annealing.

The results of optical measurements have shown that CdS thin films deposited at  $100^{\circ}$ C have 2.39eVbangap and 2.42eV bandgap have films deposited at  $200^{\circ}$ C and  $300^{\circ}$ C. This value of bandgap (2.42eV) is equal to the bandgap of CdS bulk crystal. These values of bandgap after annealing slightly decreased from 2.39eV to 2.38eV and from 2.42eV to 2.40eV.

As showed the AFM images and profiler measurements the roughness of the CdS films is 5-8nm and weakly depends on both substrate temperature and annealing.

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