# PREPARATION OF Ho-TiO<sub>2</sub> BASED PHOTOANODE FOR IMPROVED ENERGY CONVERSION EFFICIENCY OF DYE SENSITIZED SOLAR CELL

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Holmium doped titanium dioxide (Ho-TiO<sub>2</sub>) nanoparticles synthesized by sol-gel method were used to fabricate photoanode for dve sensitized solar cell (DSSC). Powder X-ray diffraction (PXRD) technique was used to confirm the single phase crystalline structure of nanoparticles, UV-Vis absorption spectroscopy was used to probe the optical properties whereas scanning electron microscope (SEM) showed the surface morphology and average particle size of the nanoparticles. Photocurrent density-voltage (J-V)characteristics measured under simulated solar light (AM 1.5G, 100 mW cm<sup>-2</sup>) showed that DSSCs based on Ho doped TiO<sub>2</sub> photoanode have fairly higher power conversion efficiency (PCE) in comparison with undoped TiO<sub>2</sub> based DSSC. Namely the DSSC with optimum 1 wt.% Ho doping exhibited a PCE ( $\eta$ ) of 4.89% (short-circuit current density (Jsc) 13.49 mA cm<sup>-2</sup>, open-circuir current density (Voc) 0.57 V and fill factor (FF) 0.54) which is 23.48% higher than undoped TiO<sub>2</sub> based DSSC having PCE of 3.96% (Jsc 10.98 mA cm<sup>-2</sup>, Voc 0.59 V and FF 0.54) under similar conditions. Enhancement in power conversion efficiency (PCE) was attributed to narrowed bandgap of Ho doped TiO<sub>2</sub> photoanode which expands response in visible region to effectively utilize sunlight and acceleration in charge transport by suppressing charge transport resistance.

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# **1. Introduction**

Energy consumption and demand is increasing globally due to increasing world population at an alarming rate. As the development of modern society is based on the availability of uninterrupted supply of energy, therefore, it is need of time to explore renewable sources of energy [1-6]. Solar energy is the most abundant source of energy which can be converted to electrical energy through photovoltaic devices [7, 8]. Among solar devices dye sensitized solar cells (DSSC) have got considerable attention due to easy availability of its materials, economical, lightweight and its high power conversion efficiency in diffused light [9-14]. A DSSC is composed of a semiconductor photoanode sensitized by adsorbed dye molecules, redox electrolyte and a platinum counter electrode [15-18]. In a DSSC photoanode is mainly responsible for the photocurrent density and the efficiency of device as it transports photo-inducded electrons from excited dye molecules adsorbed on its surface [10, 19, 20]. Among photoanode materials anatase TiO<sub>2</sub> is the most investigated material which has given excellent photostability, low toxicity and superior device efficiency [10, 13, 21, 22]. As it absorbs mostly in UV region which is only 5% of solar spectrum due to its wide bandgap (3.2 eV) limits its large scale use. And numerous dyes cannot inject electrons into conduction band (CB) of TiO<sub>2</sub> due to insufficient driving force which can be expressed as the difference between excited state of dye molecules and conduction band (CB) of TiO<sub>2</sub>. Therefore, in order to enhance energy harvesting efficiency of DSSC device alternative

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photoanodes that should exhibit better photoresponse under irradiation of visible light are urgently needed [23-26].

DSSCs based on nanocrystalline TiO<sub>2</sub> photoanode have got huge attention since its breakthrough study by Gratzel in 1991 [13]. To improve the efficiency of DSSC various modifications have been reported like surface modification of  $TiO_2$  photoanode by  $TiCl_4$  (which increases photocurrent by enhancing lifetime of electrons) [27], combining  $TiO_2$  with other semiconductors (SnO<sub>2</sub>, ZnO, ZrO<sub>2</sub>, Nb<sub>2</sub>O<sub>5</sub>, In<sub>2</sub>O<sub>3</sub>, WO<sub>3</sub> and La<sub>2</sub>O<sub>3</sub>) [15, 28-30], doping metals (Cu, Zn. Cr, Mg, Nb, Ta) [31-36] and nonmetals (N, S, F, I) [37-40]. Doping of TiO<sub>2</sub> with nonmetals reduces bandgap of semiconductor, improves visible light response and enhances DSSC efficiency. For example Guo et el. doped TiO<sub>2</sub> with varying amount of nitrogen and have got maximum 46% improvement in efficiency with 0.4% N doping as compared to undoped  $TiO_2$ based DSSC [41]. Similarly Yang and co-workers applied F doped TiO<sub>2</sub> photoanode to DSSC and obtained 11.3% higher device efficiency [42]. However doping of nonmetals is limited due to their electronegativity and small size [43]. Doping of metals to photoanode enhances photocurrent and reduces the electron recombination in DSSC. Zhou et al. [31] enhanced the visible light properties of TiO<sub>2</sub> by Cu doping and got almost 20 % higher DSSC performance by improving short circuit current density. Tong et al. [44] introduced intermediate bands in the bandgap of TiO<sub>2</sub> through W doping and enhanced DSSC efficiency from 6.64 % to 7.42 %. Neetu et al. [45] narrowed bandgap of TiO<sub>2</sub> photoanode by aluminum doping and got pretty high IPCE and overall efficiency of DSSC. Li et al. [46] decorated Au on urchin like  $TiO_2$  microspheres by one step hydrothermal method. DSSC based on this Au-TiO<sub>2</sub> microspheres have given pretty high efficiency (7.21%) in comparison with only urchin like TiO<sub>2</sub> microspheres ( $\eta$ =6.10%) and pure P25 ( $\eta$ =5.19%) based DSSCs. Some researchers modified TiO<sub>2</sub> photoanode by co-doping with different metals or nonmetals and got the benefit of synergic effect of co-doping for enhanced photovoltaic performance. Park et al. [27] synthesized highly visible light responsive Zr/N-TiO<sub>2</sub> nanoparticles by sol-gel method and got excellent DSSC efficiency (12.62%). Tanyi et al. [47] enhanced efficiency of DSSC by co-doping TiO<sub>2</sub> with La and Mg. Liu et al. [48] co-doped TiO<sub>2</sub> with Zn and Mg and got excellent photovoltaic results. Majority of researchers focused on doping TiO<sub>2</sub> with main group and transition metals whereas very few groups introduced f-block elements into  $TiO_2$  photoanode. Doping of  $TiO_2$  with rare-earth elements could be very beneficial because 4f electrons can change optical, photo-physical and chemical properties due to possibility of electronic transitions in 4f orbital [49]. Zhang et al. [50] modified TiO<sub>2</sub> electrode with cerium doping and got excellent photovoltaic response by tuning bandgap positively. Yao et al. [51] had harvested 33.3% higher DSSC efficiency by modifying photoanode with  $Nd-TiO_2$  nanorods. Improvement in efficiency was due to better electron injection properties as well as due to reduced electron transfer resistance. Similarly Wei et al. and Qin et al. enhance visible light response of photoelectrode by introducing Sm into TiO<sub>2</sub> nanocrystals and got excellent DSSC performance [49, 52].

In present work, Ho doped  $TiO_2$  nanoparticles were prepared by modified sol-gel technique and structural, optical and morphological characterizations were carried out by X-ray diffraction, UV-vis absorption spectroscopy and scanning electron microscopy (SEM) respectively. Photoanodes were fabricated from synthesized materials and photovoltaic properties of DSSCs were probed through current-density vs voltage (*J-V*) and incident photon to current conversion efficiency (IPCE) while charge transport across the DSSC was examined by electrochemical impedance spectroscopic (EIS) analysis technique.

# 2. Experimental

### 2.1. Synthesis

All the chemicals used in this work were of analytical reagent grade and were used as received without further purification. Ho doped  $TiO_2$  nanoparticles with different Ho content (0.5, 1 and 2 wt.%) were synthesized by sol-gel method. In a typical procedure solution (1) was prepared by slowly adding 0.1mol of titanium isopropoxide into 100 mL anhydrous ethanol at room temperature under vigorous stirring. A second liquid medium (2) was prepared by mixing 30 mL ethanol and 10 mL acetic acid. Appropriate amount of Ho(NO<sub>3</sub>)<sub>3</sub>. 5H<sub>2</sub>O was dissolved into

#### 2.2. Preparation of electrodes and DSSC fabrication

The nanocrystalline TiO<sub>2</sub> photo-anode and platinum counter electrodes were fabricated on fluorine doped tin oxide (FTO) glass substrates using doctor blade technique. Glass substrates were first sonicated in detergent solution for 15 minutes and then rinsed with distilled water, acetone and ethanol respectively and finally dried in oven. To prepare printing paste for photoanode, 5g Ho-TiO<sub>2</sub> or un-doped TiO<sub>2</sub> nanoparticles powder was added into 20 mL absolute ethanol, followed by addition of 3 mL acetic acid and 1.5 mL Triton x-100. The resultant mixture was sonicated for 15 min in ultrasonic bath. Then it was vigorously mixed in mortar and pestle until a homogeneous paste formed which was concentrated at 40 °C to achieve viscous paste. It was printed on cleaned FTO substrates and annealed in furnace at 450 °C for 30 minutes. When cooled down to room temperature, these printed electrodes were dipped into N719 dye solution (0.3 mM) prepared in a mixture of tert-butanol, DMSO and acetonitrile (1:1:1) in dark for 24 hours. These photoanodes were taken out, rinsed with ethanol and dried at room temperature.

FTO glass substrates for counter electrodes were cleaned using same procedure for photoanode substrates. Platinum paste (Solaronix, Platisol T/SP) was coated on conducting side by doctor blade procedure and electrodes were annealed in furnace at 450 °C for 10 minutes.

DSSCs were assembled from these dye adsorbed working electrodes and Pt printed counter electrode in a sandwich type fashion with printed sides facing each other. Electrodes were hold together by binding clips, electrolyte (solaronix, iodolyte AN-50) was injected between the electrodes with the help of a syringe and binder clips were loosed alternatively so that electrolyte cover entire active cell area.

#### **2.3.** Characterization

under same conditions.

Crystalline phase of undoped and Ho doped  $\text{TiO}_2$  nanoparticles was measured using PANAnalytical X-ray diffractometer with Cu-Ka radiation. UV-vis absorption studies were performed using spectrophotometer (Shimadzu UV-1601, japan) in the range of 200-800 nm. Surface morphologies of prepared nanoparticles were observed with scanning electron microscopy (SEM) Jeol JSM-6510LV. Photocurrent-voltage (*J-V*) measurements of DSSC devices were done on a Keithley 2400 source meter under illumination by solar simulator (SOL3A, Oreal) at AM 1.5 G illumination with light intensities of 100 mW cm<sup>-2</sup>. Incident photon to current conversion efficiency (IPCE) properties were studied using QEX10, PV Measurements system. Electrochemical impedance spectroscopic (EIS) studies were carried out under illumination (with xenon lamp of 450 W) at 100 mW cm<sup>-2</sup> using solar simulator (Oriel SOL3A,) set at bias voltage of Voc. Measurements were obtained from 100 mHz to 200 kHz and Bio-Logic, Z-fit software was used to fit spectras.

# 3. Results and discussion

Structural characterization of Ho doped TiO<sub>2</sub> and pure TiO<sub>2</sub> nanoparticles were carried out by X-ray diffraction measurements. Figure 1 displays the XRD pattern of undoped TiO<sub>2</sub> and 0.5%, 1% and 2 wt.% Ho doped TiO<sub>2</sub> nanoparticles. XRD results indicate that all the samples have pure anatase structure (matched with standard JCPDS card no 21-1272), having no individual holmium oxide peak, suggesting that Ho ions are precisely doped into TiO<sub>2</sub> lattice. Ho doped samples have relatively lower peak intensities than undoped TiO<sub>2</sub>. Crystallite size of nanoparticles was estimated with Scherrer equation [53] from full width at half maximum for major diffraction peak (101) which reveals that Ho doping cause a decrease in particle size. Due to insertion of Ho into TiO<sub>2</sub> lattice diffraction peaks can shift towards lower or higher angle [54], as can be seen in Fig. 1. In Ho doped samples Peak broadening and shifting towards lower diffraction angle at (101), (004), (200) and (105) have been observed as compared to un-doped  $TiO_2$  which confirms successful doping.



Fig. 1. XRD pattern of synthesized TiO<sub>2</sub> and Ho-TiO<sub>2</sub> nanoparticles.

Surface morphology of the undoped and Ho doped  $\text{TiO}_2$  particles was determined with scanning electron microscopy (Figure 2) and particles sizes were estimated with imageJ software. Pure TiO<sub>2</sub> nanoparticles have cubical shape with wide size distribution ranging from 30-70 nm. Cubical shape retained in 0.5 wt.% Ho doped TiO<sub>2</sub> nanoparticles while particle size reduced (nearly 28-55 nm), whereas 1 wt.% Ho doped TiO<sub>2</sub> particles are of elongated and some irregular shapes and is nearly 25-50 nm. In 2 wt.% Ho doped TiO<sub>2</sub> sample, particles with little agglomeration are of different shapes and size is nearly 30-55 nm. SEM images clearly shows that upon Ho doping into TiO<sub>2</sub>, particles size decreases thus providing higher specific surface area which is beneficial for sufficient dye loading on photoanode.



Fig. 2. SEM images of synthesized TiO<sub>2</sub> (a) and Ho-TiO<sub>2</sub> (b-d) nanoparticles.

To study the effect of Ho doping on optical properties of  $TiO_2$  UV-vis absorption studies have been carried out of all synthesized samples. The absorption spectra of undoped  $TiO_2$  and Ho doped  $TiO_2$  nanoparticles are given in figure 3 which indicate that on Ho doping absorption increased and absorption window widened towards longer wavelength which is beneficial for DSSC photoanode material. In undoped  $TiO_2$  sample, absorption occurs due to electronic transitions from O-2p to Ti-3d orbital while in doped samples it is Ho-4f orbital electrons which induced red shift in absorption spectrum [55]. This means Ho doping cause the narrowing of band gap of  $TiO_2$  nanoparticles as calculated by plotting  $(ahv)^{1/2}$  vs hv photon energy shown in Fig. 3 in inset. Comparing 0.5 wt.%, 1 wt.% and 2 wt.% Ho doped samples it is observed absorption increases from 0.5-1 wt.% and decreases from 1-2 wt.% doping level.



Fig. 3. UV-vis absorption spectra and in inset  $(ahv)^{1/2}$  vs hv of TiO<sub>2</sub> and Ho-TiO<sub>2</sub> nanoparticles.

Table 1. Crystallite size and bandgap of synthesized TiO<sub>2</sub> and different Ho-TiO<sub>2</sub> nanoparticles.

Sample	*Crystallite size (nm)	*Bandgap (eV)
Undoped TiO <sub>2</sub>	20.60	3.17
0.5 wt.% Ho-TiO <sub>2</sub>	15.70	2.95
1 wt.% Ho-TiO <sub>2</sub>	13.20	2.86
2 wt.% Ho-TiO <sub>2</sub>	16.32	3.02

<sup>\*</sup>Crystallite size was calculated using Scherrer equation from major diffraction peak (101) while bandgap was calculated from UV-vis data by plotting  $(\alpha hv)^{1/2}$  vs hv photon energy.

Dye adsorption capability of photoanodes were determined by measuring UV-vis absorption spectra of dye adsorbed photoanodes. Fig. 4 shows that Ho doped  $TiO_2$  based photoanodes have higher absorption than undoped  $TiO_2$  based photoanode. Thus it is confirmed that Ho doping into  $TiO_2$  enhances its dye adsorption ability and expands absorption in visible region. Higher dye adsorption facilitate wide range of solar light capturing and thus producing more electrons by photoanode which ultimately lead to better DSSC efficiency.



Fig. 4. UV-Vis absorption spectra of dye (N719) sensitized photoanodes.

To explore the effect of Ho doping on photovoltaic properties of TiO<sub>2</sub>, DSSCs were fabricated from un-doped  $TiO_2$  and Ho doped  $TiO_2$  nanoparticles. Current density-voltage (J-V) characteristics of DSSCs measured under standard illumination conditions (AM 1.5G, 100  $mW/cm^{2}$ ) are given in Figure 5 while detailed photovoltaic parameters are given in table 2. Results showed that DSSCs based on Ho doped photoanodes (0.5 wt%, 1 wt% and 2 wt%) have higher device efficiency compared to undoped TiO<sub>2</sub> based photoanode. DSSC efficiency increases on increasing Ho content in photoanode up to 1 wt.% doping and then decreases on further increasing holmium quantity. DSSC with 1 wt% Ho-TiO<sub>2</sub> have given maximum energy harvesting efficiency ( $\eta$ ) of 4.89% with Jsc of 13.49 mA cm<sup>-2</sup>, Voc 0.57 V, FF 0.54. This is 23.48% higher efficiency ( $\eta$ ) than un-doped TiO<sub>2</sub> based DSSC which have overall efficiency ( $\eta$ ) of 3.96%. Photovoltaic parameters indicate that Ho doped DSSCs have higher Jsc and overall efficiency  $(\eta)$  but no significant effect on Voc and fill factor of devices, it means Ho doping increases electron density in TiO<sub>2</sub> and promote electron transport properties by reducing electron transfer resistance. Improvement in efficiency can be ascribed to enhanced absorption which favors more effectively consuming UV-Visible light, increment in Jsc arising from efficient electron injection, reduced charge transport resistance and translates to higher PCE [56, 57]. A downfall in efficiency at 2 wt.% Ho doping was possibly due to formation of defects in crystal lattice which trap the photogenerated electrons, thus lowering the Jsc and hence the whole efficiency of device [31].



Fig. 5. J-V curves of DSSCs fabricated from TiO<sub>2</sub> and Ho-TiO<sub>2</sub> nanoparticles.

DSSC	$Jsc (mA cm^{-2})$	Voc (V)	FF	PCE (%)
Undoped TiO <sub>2</sub>	10.98	0.59	0.54	3.96
0.5 wt.% Ho-TiO <sub>2</sub>	12.30	0.57	0.53	4.21
1 wt.% Ho-TiO <sub>2</sub>	13.49	0.57	0.54	4.89
2 wt.% Ho-TiO <sub>2</sub>	11.92	0.58	0.53	4.07

Table 2. Photovoltaic properties of synthesized  $TiO_2$  and different Ho-TiO<sub>2</sub> based DSSCs.

Incident photon to current conversion efficiency (IPCE) measurements: IPCE of device is the ratio between total number of electrons produced by light in external circuit and incident photons number. Fig. 6 displays the IPCE spectra of un-doped and Ho doped TiO<sub>2</sub> based DSSCs, measured between 300-900 nm. It shows that Ho doped devices have fairly higher IPCE compared to pure TiO<sub>2</sub> based device in entire region. At 540 nm maximum IPCE values observed for Ho doped TiO<sub>2</sub> (0.5%, 1%, 2%) based DSSC were 52%, 55.40% and 49.5% respectively whereas at this wavelength un-doped TiO<sub>2</sub> based DSSC has 47.62%. The higher IPCE values of Ho doped devices can be attributed to increased dye adsorption, higher absorption in visible region and efficient charge transport properties.



Fig. 6. IPCE profile of synthesized TiO<sub>2</sub> and Ho-TiO<sub>2</sub> based DSSCs.

Electrochemical impedance spectroscopy (EIS) is an important spectroscopic method to study charge transport in DSSCs [44]. To investigate the change in electron transport in DSSC upon doping  $TiO_2$  with Ho electrochemical impedance spectroscopic analysis have been carried out. In EIS Nyquist plot generally two semicircles are found, small semicircle represent electron transport resistance at the interface of FTO/TiO<sub>2</sub> photoanode and electrolyte/Pt counter electrode. Bigger semicircle in low frequency region represents the charge transport across interface of mesoporous photoanode/dye/electrolyte [44, 58, 59].



Fig. 7. Nyquist plot of DSSCs measured under standard solar irradiation (AM 1.5G).

EIS spectra of DSSCs based on un-doped TiO<sub>2</sub> and Ho doped TiO<sub>2</sub> are presented in Figure 7. In these spectra it can be seen that larger semicircle is decreasing upon Ho doping into TiO<sub>2</sub> which means acceleration in electron transport is taking place due to decrease in charge transport resistance resulting in increased *Jsc* and efficiency of cells. This effect is more pronounced in DSSCs with 0.5 wt.% and 1 wt.% Ho doping and when doping amount increased to 2 wt.% semicircle size again increases resulting a decrease in *Jsc* and efficiency of cell. Thus EIS studies inferred that DSSC based on 1 wt.% Ho doping in TiO<sub>2</sub> has lowest charge transport resistance, higher Jsc and hence higher efficiency of device.

## 4. Conclusion

In summary Ho-TiO<sub>2</sub> nanoparticles prepared by sol-gel method were successfully utilized to fabricate photoanode for N719 sensitized DSSCs. X-ray diffractograms confirm the successful incorporation of Ho into TiO<sub>2</sub> anatase lattice as there were no individual holmium oxide peaks. UV-vis absorption measurements showed red shift in absorption thus narrowing the bandgap which is useful for efficient utilization of sunlight. Photovoltaic properties of DSSCs were greatly enhanced by Ho doping into TiO<sub>2</sub>. DSSC with 1 wt % Ho-TiO<sub>2</sub> yielded best PCE ( $\eta$ ) 4.89% while under similar conditions of one sun illumination AM 1.5 undoped TiO<sub>2</sub> based DSSC have given PCE ( $\eta$ ) of 3.96%. IPCE measurements followed the same pattern of J-V measurements, i.e. 1 wt% Ho- TiO<sub>2</sub> exhibited maximum value of 55.40% at  $\lambda_{max}$  540 nm in comparison with undoped TiO<sub>2</sub> which have resulted a maximum of 47.62%. EIS studies revealed that electron transport increases after Ho doping by suppressing charge transport resistance. Hence Ho doping into TiO<sub>2</sub> made it a better option as a photoanode for DSSC in comparison with pure TiO<sub>2</sub>.

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